
Modern Chemistry Pupil S Addition

Holt Modern Chemistry

Modern Chemistry with Its Practical Applications (Classic Reprint)

Ontario High School Chemistry

Holt Chemistry

Teaching Chemistry - A Studybook

Student Solutions Manual for Oxtoby, Gillis, and Campion's Principles of Modern Chemistry

Hmh Modern Chemistry Florida

Elements of Modern Chemistry (Classic Reprint)

Principles of Modern Chemistry

From Classical to modern chemistry

A Treatise on Pharmacy. Designed as a Text-book for the Student ... Containing the Official and Many Unofficial Formulas and Numerous Examples of Extemporaneous Prescriptions ... Third Edition, Thoroughly Revised ... with Important Additions. With ... Illustrations

The Medical Times and Gazette

John Dalton and the Rise of Modern Chemistry

Elementary Modern Chemistry (Classic Reprint)
The London student
The Educational Times, and Journal of the College of Preceptors
Knowledge
Relevant Chemistry Education
Modern Quantum Chemistry
Student Solutions Manual for Oxtoby/Gillis/Butler's Principles of Modern Chemistry,
8th
Elements of Modern Chemistry
Concepts and Methods in Modern Theoretical Chemistry
Pharmaceutical Journal
Experimental Exercises and Problems in Elementary Chemistry
Principles of Modern Chemistry
Pharmaceutical Journal;
General Chemistry
Frankenstein (Second International Student Edition) (Norton Critical Editions)
Physical Chemistry
The American Journal of the Medical Sciences
Education Outlook
Fundamentals of Chemistry

Holt McDougal Modern Chemistry
Elementary Modern Chemistry
Educational Times
Medical Examiner
Lessons in Elementary Chemistry
Modern Chemistry
The London and Edinburgh Monthly Journal of Medical Science
Congressional Record

Modern Chemistry Pupil S Addition <http://connect.hi.u.edu> *by guest*

DAVIES COLBY

Holt Modern Chemistry Forgotten Books

You cannot expect to succeed in a serious chemistry course without solving problems, which are universally used to illustrate concepts and to test understanding. This manual offers

detailed solutions to all of the odd-numbered problems in the text to assist you in working through them.

Modern Chemistry with Its Practical Applications (Classic Reprint)

Forgotten Books

Concepts and Methods in Modern Theoretical Chemistry: Statistical Mechanics, the second book in a two-volume set, focuses on the dynamics of systems and phenomena. A new addition

to the series Atoms, Molecules, and Clusters, this book offers chapters written by experts in their fields. It enables readers to learn how concepts from ab initio quantum chemistry and density functional theory (DFT) can be used to describe, understand, and predict chemical dynamics. This book covers a wide range of subjects, including discussions on the following topics: Time-dependent DFT Quantum fluid dynamics (QFD) Photodynamic control, nonlinear dynamics, and quantum hydrodynamics Molecules in a laser field, charge carrier mobility, and excitation energy transfer Mechanisms of chemical reactions Nucleation, quantum Brownian motion, and the third law of thermodynamics Transport properties of binary mixtures Although most of the

chapters are written at a level that is accessible to a senior graduate student, experienced researchers will also find interesting new insights in these experts' perspectives. This book provides an invaluable resource toward understanding the whole gamut of atoms, molecules, and clusters.

Ontario High School Chemistry CRC Press

This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1917 edition. Excerpt: ... (6) Columns for Discount on Purchases and Discount on Notes on the same side of the Cash Book; (c) Columns for Discount on Sales and Cash Sales on the debit

side of the Cash Book; (d) Departmental columns in the Sales Book and in the Purchase Book. Controlling Accounts.-- The addition of special columns in books of original entry makes possible the keeping of Controlling Accounts. The most common examples of such accounts are Accounts Receivable account and Accounts Payable account. These summary accounts, respectively, displace individual customers' and creditors' accounts in the Ledger. The customers' accounts are then segregated in another book called the Sales Ledger or Customers' Ledger, while the creditors' accounts are kept in the Purchase or Creditors' Ledger. The original Ledger, now much reduced in size, is called the General Ledger. The Trial Balance now refers to the accounts

in the General Ledger. It is evident that the task of taking a Trial Balance is greatly simplified because so many fewer accounts are involved. A Schedule of Accounts Receivable is then prepared, consisting of the balances found in the Sales Ledger, and its total must agree with the balance of the Accounts Receivable account shown in the Trial Balance. A similar Schedule of Accounts Payable, made up of all the balances in the Purchase Ledger, is prepared, and it must agree with the balance of the Accounts Payable account of the General Ledger." The Balance Sheet.--In the more elementary part of the text, the student learned how to prepare a Statement of Assets and Liabilities for the purpose of disclosing the net capital of an enterprise. In the present chapter

he was shown how to prepare a similar statement, the Balance Sheet. For all practical...

Holt Chemistry W. W. Norton & Company

This book is aimed at chemistry teachers, teacher educators, chemistry education researchers, and all those who are interested in increasing the relevance of chemistry teaching and learning as well as students' perception of it. The book consists of 20 chapters. Each chapter focuses on a certain issue related to the relevance of chemistry education. These chapters are based on a recently suggested model of the relevance of science education, encompassing individual, societal, and vocational relevance, its present and future implications, as well as its intrinsic

and extrinsic aspects. "Two highly distinguished chemical educators, Ingo Eilks and AviHofstein, have brought together 40 internationally renowned colleagues from 16 countries to offer an authoritative view of chemistry teaching today. Between them, the authors, in 20 chapters, give an exceptional description of the current state of chemical education and signpost the future in both research and in the classroom. There is special emphasis on the many attempts to enthuse students with an understanding of the central science, chemistry, which will be helped by having an appreciation of the role of the science in today's world. Themes which transcend all education such as collaborative work, communication skills, attitudes, inquiry learning and teaching,

and problem solving are covered in detail and used in the context of teaching modern chemistry. The book is divided into four parts which describe the individual, the societal, the vocational and economic, and the non-formal dimensions and the editors bring all the disparate leads into a coherent narrative, that will be highly satisfying to experienced and new researchers and to teachers with the daunting task of teaching such an intellectually demanding subject. Just a brief glance at the index and the references will convince anyone interested in chemical education that this book is well worth studying; it is scholarly and readable and has tackled the most important issues in chemical education today and in the foreseeable future.” – Professor David

Waddington, Emeritus Professor in Chemistry Education, University of York, United Kingdom

Teaching Chemistry – A Studybook
Springer

Excerpt from Elementary Modern Chemistry The beginner in the study of chemistry finds interest and pleasure, first of all, in the remarkable new facts which are placed before him, and this is both natural and desirable. No one can master the science without much study of properties and reactions, and minute acquaintance with as many phenomena as possible is a fundamental requisite. Modern chemistry has, however, a number of general laws which appear everywhere through it, connecting various facts in the most interesting way. Before study has gone very far it

becomes necessary to begin the statement of these laws and to indicate how the facts already learned group themselves about them. Two purposes are served by this method of presentation. First, a real science of chemistry begins to unfold itself to the student, and second, the acquisition of more new facts is made easier and more interesting. We have planned in this book to present a sufficient number of facts and experiments to fill the time usually devoted to a first course, and at the same time we have endeavored to fit these facts, as far as possible, to the simpler of the general laws now firmly established as the basis of the science of chemistry. It is not to be expected or desired that every teacher will wish to follow exactly the order we have chosen.

About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works.

[Student Solutions Manual for Oxtoby, Gillis, and Campion's Principles of](#)

Modern Chemistry Courier Corporation

This graduate-level text explains the modern in-depth approaches to the calculation of electronic structure and the properties of molecules. Largely self-contained, it features more than 150 exercises. 1989 edition.

Hmh Modern Chemistry Florida

Academic Press

Long considered the standard for honors and high-level mainstream general chemistry courses, PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern, rigorous, and chemically and mathematically accurate text on the market. This authoritative text features an "atoms first" approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure

(Chapter 6), Electrochemistry (Chapter 17), and Molecular Spectroscopy and Photochemistry (Chapter 20). In addition, the text utilizes mathematically accurate and artistic atomic and molecular orbital art, and is student friendly without compromising its rigor. End-of-chapter study aids focus on only the most important key objectives, equations and concepts, making it easier for students to locate chapter content, while applications to a wide range of disciplines, such as biology, chemical engineering, biochemistry, and medicine deepen students' understanding of the relevance of chemistry beyond the classroom.

Elements of Modern Chemistry (Classic Reprint) Cengage AU

Fundamentals of Chemistry, Fourth

Edition covers the fundamentals of chemistry. The book describes the formation of ionic and covalent bonds; the Lewis theory of bonding; resonance; and the shape of molecules. The book then discusses the theory and some applications of the four kinds of spectroscopy: ultraviolet, infrared, nuclear (proton) magnetic resonance, and mass. Topics that combine environmental significance with descriptive chemistry, including atmospheric pollution from automobile exhaust; the metallurgy of iron and aluminum; corrosion; reactions involving ozone in the upper atmosphere; and the methods of controlling the pollution of air and water, are also considered. Chemists and students taking courses related to chemistry and environmental

chemistry will find the book invaluable. *Principles of Modern Chemistry*
Forgotten Books
Excerpt from Elementary Modern Chemistry The beginner in the study of chemistry finds interest and pleasure, first of all, in the remarkable new facts which are placed before him, and this is both natural and desirable. No one can master the science without much study of proper ties and reactions, and minute acquaintance with as many phenomena as possible is a fundamental requisite. Modern chemistry has, however, a number of general laws which appear. Everywhere through it, connecting various facts in the most interesting way. Before study has gone very far it becomes necessary to begin the statement of these laws and to indicate

how the facts already learned group themselves about them. Two purposes are served by this method of presentation. First, a real science of chemistry begins to unfold itself to the student, and second, the acquisition of more new facts is made easier and more interesting. We have planned in this book to present a sufficient number of facts and experiments to fill the time usually devoted to a first course, and at the same time we have endeavored to fit these facts, as far as possible, to the simpler of the general laws now firmly established as the basis of the science of chemistry. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical

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From Classical to modern chemistry CUP Archive

Excerpt from Lessons in Elementary Chemistry: Inorganic and Organic In the following pages I have endeavoured to arrange the most important facts and principles of Modern Chemistry in a plain

but precise and scientific form, suited to the present requirements of Elementary Instruction. For the purpose of facilitating the attainment of that exactitude in the knowledge of the subject, without which the introduction of physical science into the school system is worse than useless, I have added a series of Exercises and Questions upon the Lessons. The pupil must learn to work out accurately both the numerical and descriptive examples, and the teacher may find it advisable to add largely to their number. Particular attention should be given to the calculation of the relations between the weights of gases, and their volumes measured under varying circumstances of temperature and pressure. The metric system of weights and measures and the

centigrade thermometric scale are used throughout the work. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works.

A Treatise on Pharmacy. Designed as a

Text-book for the Student ... Containing the Official and Many Unofficial Formulas and Numerous Examples of Extemporaneous Prescriptions ... Third Edition, Thoroughly Revised ... with Important Additions. With ... Illustrations
Springer Science & Business Media
Designed for a two-semester introductory course sequence in physical chemistry, *Physical Chemistry: A Modern Introduction*, Second Edition offers a streamlined introduction to the subject. Focusing on core concepts, the text stresses fundamental issues and includes basic examples rather than the myriad of applications often presented in other, more encyclopedic books. Physical chemistry need not appear as a large assortment of different, disconnected, and sometimes intimidating topics.

Instead, students should see that physical chemistry provides a coherent framework for chemical knowledge, from the molecular to the macroscopic level. The book offers: Novel organization to foster student understanding, giving students the strongest sophistication in the least amount of time and preparing them to tackle more challenging topics Strong problem-solving emphasis, with numerous end-of-chapter practice exercises, over two dozen in-text worked examples, and a number of clearly identified spreadsheet exercises A quick review in calculus, via an appendix providing the necessary mathematical background for the study of physical chemistry Powerful streamlined development of group theory and advanced topics in quantum mechanics,

via appendices covering molecular symmetry and special quantum mechanical approaches

The Medical Times and Gazette CRC Press

This book focuses on developing and updating prospective and practicing chemistry teachers' pedagogical content knowledge. The 11 chapters of the book discuss the most essential theories from general and science education, and in the second part of each of the chapters apply the theory to examples from the chemistry classroom. Key sentences, tasks for self-assessment, and suggestions for further reading are also included. The book is focused on many different issues a teacher of chemistry is concerned with. The chapters provide contemporary discussions of the

chemistry curriculum, objectives and assessment, motivation, learning difficulties, linguistic issues, practical work, student active pedagogies, ICT, informal learning, continuous professional development, and teaching chemistry in developing environments. This book, with contributions from many of the world's top experts in chemistry education, is a major publication offering something that has not previously been available. Within this single volume, chemistry teachers, teacher educators, and prospective teachers will find information and advice relating to key issues in teaching (such as the curriculum, assessment and so forth), but contextualised in terms of the specifics of teaching and learning of chemistry, and drawing upon the

extensive research in the field. Moreover, the book is written in a scholarly style with extensive citations to the literature, thus providing an excellent starting point for teachers and research students undertaking scholarly studies in chemistry education; whilst, at the same time, offering insight and practical advice to support the planning of effective chemistry teaching. This book should be considered essential reading for those preparing for chemistry teaching, and will be an important addition to the libraries of all concerned with chemical education. Dr Keith S. Taber (University of Cambridge; Editor: Chemistry Education Research and Practice) The highly regarded collection of authors in this book fills a critical void by providing an essential resource for

teachers of chemistry to enhance pedagogical content knowledge for teaching modern chemistry. Through clever orchestration of examples and theory, and with carefully framed guiding questions, the book equips teachers to act on the relevance of essential chemistry knowledge to navigate such challenges as context, motivation to learn, thinking, activity, language, assessment, and maintaining professional expertise. If you are a secondary or post-secondary teacher of chemistry, this book will quickly become a favorite well-thumbed resource! Professor Hannah Sevian (University of Massachusetts Boston)
John Dalton and the Rise of Modern Chemistry Holt McDougal
Excerpt from Elements of Modern

Chemistry This book is translated from the fourth French edition by my pupil and friend, M. Greene, whose perfect familiarity with the French language and thorough competence, at the same time, in chemistry I have had occasion to appreciate. The translation is, then, a faithful, or even improved, representation of the original work, in which he will certainly have detected and corrected some faults. The French editions succeed each other rapidly, showing that this little book responds to an educational need. It has been the endeavor to keep it up with the current of the latest discoveries, and in it to condense a considerable number of exact and well-selected facts, without banishing the theory which binds them together. Thus, the origin and foundation

of the atomic theory have been given, as far as possible, in historical order. The notions concerning atomicity, so important for the appreciation of the structure of combinations and for the interpretation of chemical reactions, are presented in an elementary form. The reader will remark that the history of the metalloids is relatively more developed than the remainder of the book. Indeed, this is the fundamental part of chemistry, and a familiar knowledge of it is indispensable to the fruitful study of the metals and of organic chemistry. It is also the most attractive portion for beginners, for it is the most easily understood. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This

book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works.

Elementary Modern Chemistry

(Classic Reprint) Boucher Press
Long considered the standard for honors and high-level mainstream general chemistry courses, PRINCIPLES OF MODERN CHEMISTRY, 7e continues to

set the standard as the most modern, rigorous, and chemically and mathematically accurate text on the market. Thoroughly revised throughout to strengthen its sound atoms first approach, this authoritative text now features new and updated content, and more mathematically accurate and artistic atomic and molecular orbital art. In addition, the text is now more student friendly without compromising its rigor. End-of-chapter study aids now focus on only the most important key objectives, equations and concepts, making it easier for students to locate chapter content, while new applications to a wide range of disciplines, such as biology, chemical engineering, biochemistry, and medicine deepen students' understanding of the relevance of chemistry beyond the

classroom. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

The London student Forgotten Books Excerpt from Ontario High School Chemistry: Pupils Text-Book This book is published in two parts, in accordance with what is believed to be good pedagogical practice, viz.: To enable pupils to gain a knowledge of chemistry by doing chemistry, not simply by "getting it up." For this purpose it is intended that both parts of the book shall be in the hands of the scholars; and that when a topic is taken up the practical exercises relating to it shall be worked, as directed in the Manual, and the observations necessary for the lesson shall be recorded for reference.

After the notes relating to the subject have been satisfactorily completed, it is expected that the teacher shall discuss the results with the class in a systematic way, and shall guide the pupils to proper conclusions in accordance with chemical laws and principles. In this connection the descriptive part of the Text-book will come into service. Manifestly the pupils note-book will be an important part of his material outfit, for it will form the connecting link between the exercises of the Manual and the description of the Text. The Chemistry of the Middle School should serve a double purpose for the pupil who properly studies it. One of these aims at the development of that mental attitude called the scientific mind which investigates, weighs evidence, and forms judgments based on reason.

The other consists in imparting to the student such a knowledge of the facts, the principles and the methods of the science that he will be able either to acquire further information of the same kind by independent reading, or to take up a more advanced course of chemical study. Without doubt, the educational function is the more important one; and, while the two objects have some elements in common, there is yet divergence enough between them to cause hesitancy, at times, as to which of two courses is the preferable one to follow. Under such circumstances there must be some compromise between what is best and what is possible, especially when the work has to be curtailed to meet the requirements of a time limit. It has been thought best to

treat the subject in accordance with modern theory, and to introduce from the first the ideas and the terminology needed for that purpose, in order that the student may not have to discard, at a latter period, what he learns regarding chemistry in the High School. For these reasons the method of treatment and the choice of material will vary somewhat from those hitherto adopted. The present equipment of the schools, the relations of science to daily life, and the character and aims of secondary education in the province have all been considered in determining what shall be taken up in the course, and how it shall be dealt with. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This

book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works.

The Educational Times, and Journal of the College of Preceptors Forgotten Books

Excerpt from *Modern Chemistry With Its Practical Applications* IN preparing this book for use in secondary schools, I have

endeavored to look at the science from the viewpoint of the students themselves. The fault With many texts upon the same subject is that the position of the learner has been diste garded; the books have been encyclopaedic; they have pre sented a great number of facts as a skeleton or framework, but this skeleton has not been clothed with muscle and ani mated With life. N O more fascinating subject huds a place in the curricula of our secondary schools, yet to the average student chemistry is too often but an irksome task. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses

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Knowledge Forgotten Books

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appreciate. The translation is, then, a faithful, or even improved, representation of the original work, in which he will certainly have detected and corrected some faults. The French editions succeed each other rapidly, showing that this little book responds to an educational need. It has been the endeavor to keep it up with the current of the latest discoveries, and in it to condense a considerable number of exact and well-selected facts, without banishing the theory which binds them together. Thus, the origin and foundation of the atomic theory have been given, as far as possible, in historical order. The notions concerning atomicity, so important for the appreciation of the structure of combinations and for the interpretation of chemical reactions, are

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Relevant Chemistry Education

Pearson College Division

The best-selling student edition on the market, now available in a Second Edition. Almost two centuries after its publication, Frankenstein remains an indisputably classic text and Mary Shelley's finest work. This extensively revised Norton Critical Edition includes new texts and illustrative materials that convey the enduring global conversation about Frankenstein and its author. The text is that of the 1818 first edition, published in three volumes by Lackington, Hughes, Harding, Mavor, and Jones. It is accompanied by an expansive new preface, explanatory annotations, a map of Geneva and its environs, and seven illustrations, five of them new to the Second Edition. Context is provided in three supporting sections:

“Circumstance, Influence, Composition, Revision,” “Reception, Impact, Adaptation,” and “Sources, Influences, Analogues.” Among the Second Edition’s new inclusions are historical-cultural studies by Susan Tyler Hitchcock, William St. Clair, and Elizabeth Young; Chris Baldrick on the novel’s reception; and David Pirie on the novel’s many film adaptations. Related excerpts from the Bible and from John Milton’s *Paradise Lost* are now included, as is Charles Lamb’s poem “The Old Familiar Faces.” “Criticism” collects sixteen major interpretations of *Frankenstein*, nine of them new to the Second Edition. The new contributors are Peter Brooks, Bette London, Garrett Stewart, James. A. W. Heffernan, Patrick Brantlinger, Jonathan Bate, Anne Mellor, Jane Goodall, and

Christa Knellwolf. A Chronology and Selected Bibliography are also included. **Modern Quantum Chemistry** Cengage Learning

Prepare for exams and succeed in your chemistry course with this comprehensive solutions manual! Featuring worked-out solutions to every odd-numbered problem in *PRINCIPLES OF MODERN CHEMISTRY*, 8th Edition, this manual shows you how to approach and solve problems using the same step-by-step explanations found in your textbook examples.

Student Solutions Manual for Oxtoby/Gillis/Butler's Principles of Modern Chemistry, 8th Holt McDougal
Excerpt from *Experimental Exercises and Problems in Elementary Chemistry: Together With Various Chemical Tables*

and Tables for the Conversion of Common Weights and Measures Into Those of the Metric System; For the Use of Beginners The following pages were compiled expressly for the use of my own classes, and without any view to general publication. The Principals of two of our neighboring Colleges, however, on being made acquainted with the proposed plan of instruction, desiring copies for the use of their pupils, I have had an additional number printed for them; and, if the method should commend itself to teachers in other institutions, the edition could be extended to meet their wishes. The work is a mere compilation from text-books recently published in England, by some of the most eminent and successful teachers of Elementary Chemistry in the

Colleges and Universities of that country. None of the works from which the materials were extracted have, as yet, been republished in this country. They are Prof. Frankland's "Lecture Notes for Students," Prof. Galloway's "First Step in Chemistry," and "Second Step in Chemistry," Prof. Williamson's "Chemistry for Students," Prof. Roscoe's "Lessons in Elementary Chemistry," Prof. Wilson's "Inorganic Chemistry," and Prof. Hofmann's "Introduction to Modern Chemistry." Most of the Problems have been extracted from Prof. Galloway's "First Step," and if that work could have been obtained for the use of my classes, there would have been less necessity for this compilation; but repeated efforts to obtain copies, made during the last year and a half, only resulted in the

information that it was "out of print." The experience of many years, as a teacher of Chemistry, has convinced me that, even in our best Colleges, with all the aids of apparatus, and in the hands of thoroughly qualified teachers and manipulators, a large majority of pupils fail to gain such a knowledge of the principles of the science, as ought to repay them for the time faithfully spent in the attempt. This failure is due, I think, not so much to want of qualification in the teacher, or of industry in the student, as to the generally defective plan of teaching. About the Publisher Forgotten Books publishes hundreds of thousands of rare

and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works.